

**UNIVERSITY OF KANSAS**  
 Department of Physics  
 ASTR 794 — Prof. Crossfield — Spring 2025

**Problem Set 2: Transport, Distributions, Eqns of State**

**Due:** Thursday, March 6, 2025, in class

This problem set is worth **70 points**.

1. **Saha equation and pure hydrogen [15 pts].** Consider a gas of pure hydrogen at fixed density and temperature. The ionization energy of hydrogen is  $\chi_0 = 13.6$  eV. You may assume that all the hydrogen atoms (whether neutral or ionized) are in their ground energy state.

- (a) Write down the Saha equation relating the number densities of neutral and ionized hydrogen ( $n_0$  and  $n_1$ , respectively). Make reasonable approximations to use numerical values for the partition functions.

**Solution:** It's easy enough to write down the Saha equation:

$$\frac{n_1}{n_0} = \frac{2Z_1}{n_e Z_0} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT}.$$

The partition function for neutral hydrogen is

$$Z_0 = 2(1 + 2^2 e^{-\chi_0(1-1/2^2)/kT} + \dots) \approx 2 \text{ for } kT \ll \chi_0.$$

The partition function for ionized hydrogen is 1 since there are two possible orientations of the free electron's spin relative to the spin of the proton, and we've already written the factor of 2 in the Saha equation. Thus we have

$$\frac{n_1}{n_0} = \frac{1}{n_e} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT}.$$

- (b) To find the individual densities, further constraints are required. Reasonable constraints are charge neutrality ( $n_e = n_1$ ) and conservation of nucleon number ( $n_1 + n_0 = n$ ), where the total hydrogen number density  $n$  is a constant if the density  $\rho$  is fixed. Rewrite the Saha equation in terms of the hydrogen ionization fraction  $x = n_1/n$ , eliminating  $n_1$ ,  $n_0$ , and  $n_e$ . Does this equation have the expected limiting behavior for  $T \rightarrow 0$  and  $T \rightarrow \infty$ ?

**Solution:** The two constraints are the conservation of charge and nucleon number, which can be written:  $n_e = n_1$  and  $n = n_1 + n_0$ . Writing  $x = n_1/n$ , the Saha equation becomes

$$\begin{aligned} \frac{nx}{n - nx} &= \frac{1}{nx} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT} \\ \frac{x^2}{1 - x} &= \frac{1}{n} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT} \end{aligned}$$

From this equation we see that as  $T \rightarrow 0$ ,  $x \rightarrow 0$ ; i.e., no ionization occurs. And as  $T \rightarrow \infty$ ,  $x \rightarrow 1$ , indicating full ionization. These are the proper limiting behaviors.

- (c) Use the relation  $\rho = m_H n$  (where  $m_H = 1$  gm/ $N_A$ , where  $N_A = 6.023 \times 10^{23}$  is Avogadro's number) to replace  $n$  with  $\rho$ . Find an expression for the half-ionized ( $x = 0.5$ ) path in the  $\rho$ - $T$  plane. Plot this path on a log-log plot for densities in the interesting range from  $10^{-10}$ - $10^{-2}$  g cm $^{-3}$

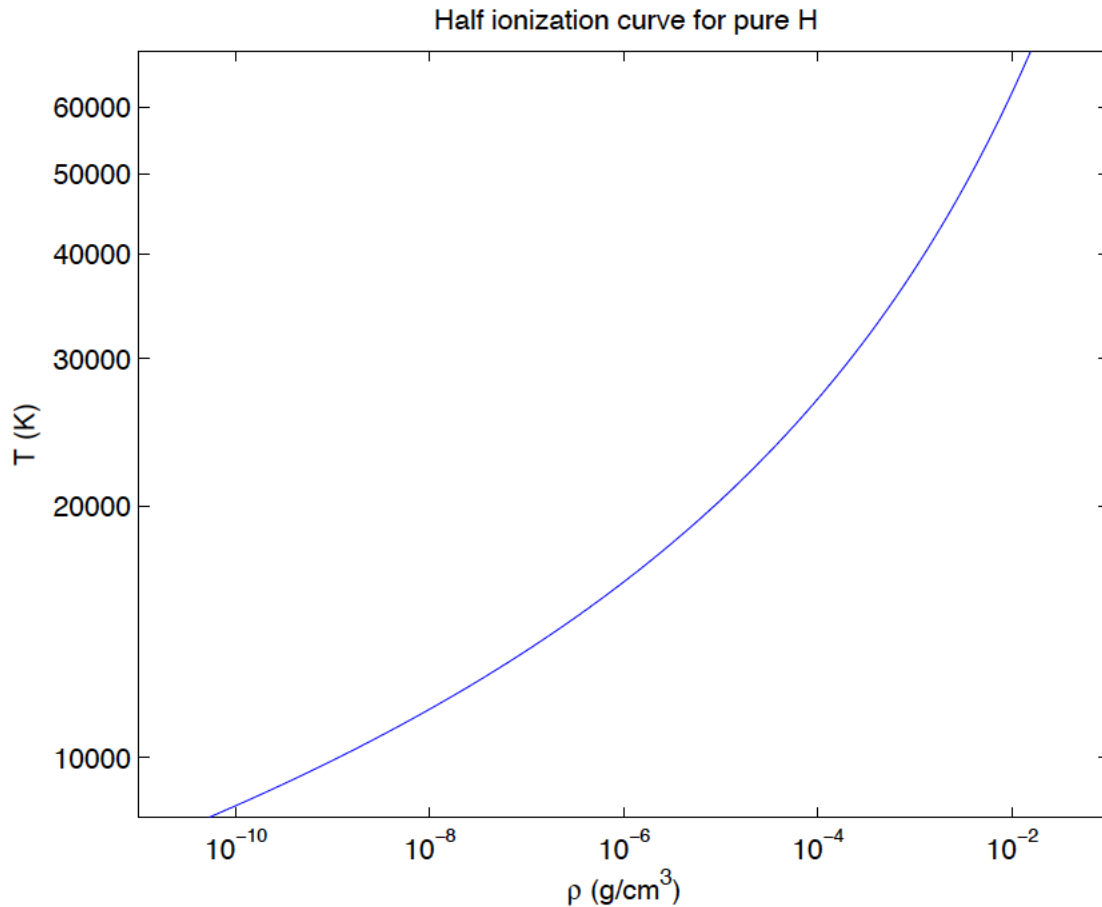


Figure 1: Half ionization curve for pure hydrogen.

**Solution:** The mass density is given by  $\rho = m_H n$ , where  $m_H$  is the mass of hydrogen,  $1/N_A$ . To get the half-ionization curve, set  $x = 0.5$  in the Saha equation to obtain

$$\rho(T) = \frac{2}{N_A} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT}.$$

This is the half-ionization curve shown in Figure 1.

## 2. Saha equation and pure helium [20 pts].

Consider a gas of pure helium at fixed density and temperature. The ionization energies for helium are  $\chi_0 = 24.6$  eV (from neutral to singly ionized) and  $\chi_1 = 54.4$  eV (from singly to doubly ionized). You may assume that all the helium atoms (whether neutral, singly ionized, or doubly ionized) are in their ground energy state. Let  $n_e$ ,  $n_0$ ,  $n_1$ , and  $n_2$  be the number densities of, respectively, free electrons, neutral atoms, singly-ionized atoms, and doubly-ionized atoms. The total number density of neutral atoms and ions is denoted by  $n$ . Define  $x_e$  as the ratio  $n_e/n$ , and let  $x_i$  be  $n_i/n$  where  $i = 0, 1, 2$ . You should assume that the gas is electrically neutral. The degeneracy factors you need for the atoms and ions are 2 for He, 4 for He<sup>+</sup>, and 2 for He<sup>2+</sup>.

- (a) Construct the ratios  $n_1/n_0$  and  $n_2/n_1$  using the Saha equation. In doing so, take care in establishing the zero points of energy for the various constituents.

**Solution:** The Saha equations are

$$\frac{n_1}{n_0} = \frac{4}{n_e} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT}$$

$$\frac{n_2}{n_1} = \frac{1}{n_e} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_1/kT}$$

- (b) Apply charge neutrality and nucleon number conservation ( $n = n_0 + n_1 + n_2$ ) and recast the above Saha equations so that only  $x_1$  and  $x_2$  appear as unknowns. The resulting two equations have  $T$  and  $n$  [or, equivalently,  $\rho = nm_{\text{He}} = n(4 \text{ gm}/N_A)$ ] as parameters.

**Solution:** Charge conservation and nucleon number conservation can be written:  $n = n_0 + n_1 + n_2$  and  $n_e = n_1 + 2n_2$ , so that the Saha equations become

$$\frac{x_1(x_1 + 2x_2)}{1 - x_1 - x_2} = \frac{4}{n} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT}$$

$$\frac{x_2(x_1 + 2x_2)}{x_1} = \frac{1}{n} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_1/kT}$$

- (c) Simultaneously solve the two Saha equations for  $x_1$  and  $x_2$  for temperatures in the range  $4 \times 10^4 \leq T \leq 2 \times 10^5$  K. Do this for a fixed density with the three values  $\rho = 10^{-4}, 10^{-6},$  or  $10^{-8} \text{ g cm}^{-3}$ . You may find it more convenient to use the logarithm of your equations. Choose a dense grid in temperature because you will soon plot the results. Once you have found  $x_1$  and  $x_2$ , also find  $x_e$  and  $x_0$  for the same range of temperature. Note that this is a numerical exercise; you will want to use a tool like Mathematica or Matlab for this.

**Solution:** The mass density is given by  $\rho = m_{\text{He}}n = 4n/N_A$ , where  $m_{\text{He}} = 4m_H = 4/N_A$  is the mass of the helium. The two Saha equations can then be written

$$f(x_1, x_2) = x_1^2 + 2x_1x_2 + \frac{16}{\rho N_A} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_0/kT} (x_1 + x_2 - 1) = 0$$

$$g(x_1, x_2) = x_1x_2 + 2x_2^2 - \frac{4}{\rho N_A} \left( \frac{2\pi m_e kT}{h^2} \right)^{3/2} e^{-\chi_1/kT} x_1 = 0$$

This set of coupled, nonlinear equations can be solved using nearly any multidimensional root-finding technique. I used a simple Newton-Raphson method, which works similarly to the Newton-Raphson method for solving a single equation. In this method, a guess is made for  $\mathbf{x} = \begin{pmatrix} x_1 \\ x_2 \end{pmatrix}$  and then the guess is refined using

$$\mathbf{x}_{\text{new}} = \mathbf{x}_{\text{old}} + \delta\mathbf{x},$$

where

$$\delta\mathbf{x} = \begin{bmatrix} \delta x_1 \\ \delta x_2 \end{bmatrix} = \begin{bmatrix} \partial f/\partial x_1 & \partial f/\partial x_2 \\ \partial g/\partial x_1 & \partial g/\partial x_2 \end{bmatrix} \begin{bmatrix} -f \\ -g \end{bmatrix}$$

Here is the MATLAB code that implements this procedure:

```
function heionization (rho, Tstart, Tspace, Tfinal)
% Plots the abundance of neutral, slightly ionized, and doubly ionized
% helium, as well as the electrons.
%
```

```

% x0 = n0/n, x1 = n1/n, x2 = n2/n, xe = ne/n, where n = rho * Na/4
%
% The temperature range is Tstart:Tspace:Tfinal % Constants

Na = 6.02214e23;
me = 9.1094e-28;
kB = 1.3807e-16;
h = 6.6261e-27;
kBeV = 8.617e-5;
chi0 = 24.6 ;
chi1 = 54.4 ;

x1 = [];
x2 = [];

%Loop through temperature for T = Tstart:Tspace:Tfinal

    % Define A and B:
    A = 16/Na/rho*(2*pi*me*kB*T/h^ 2). ^ (3/2). *exp(-chi0/kBeV./T);
    B = 4/Na/rho*(2*pi*me*kB*T/h^ 2). ^ (3/2). *exp(-chi1/kBeV./T);

    % Dumb initial guesses:
    x1guess = 0.5;
    x2guess = 0.5;
    % Calculate f and g; correct until within tolerance 0.0001;

    f = x1guess ^ 2 + 2 * x1guess * x2guess + A * (x1guess + x2guess - 1);
    g = x1guess * x2guess + 2 * x2guess ^ 2 - B * x1guess;
    err = max( abs(f), abs(g));

    while (err > 0.0001)

        M = [2*x1guess + 2*x2guess + A 2*x1guess + A; x2guess - B x1guess + 4
        * x2guess];
        dx = inv(M) * [-f -g];
        x1guess = x1guess + dx(1);
        x2guess = x2guess + dx(2);
        f = x1guess ^ 2 + 2 *x1guess *x2guess + A*(x1guess+x2guess-1);
        g = x1guess * x2guess + 2 * x2guess ^ 2- B*x1guess;
        err = max(abs(f), abs(g));
    end

    % Add solutions to list;
    x1 = [x1 x1guess];
    x2 = [x2 x2guess];

end

```

```

% Calculate xe and x0;
xe = x1 + 2 * x2;
x0 = 1 - x1 - x2 ;

Tvals = Tstart:Tspace:Tfinal ;

plot(Tvals, xe);
hold on;
plot(Tvals, x0, ':');
plot(Tvals, x1, '--');
plot(Tvals, x2, '-.');

return;

```

- (d) Plot all your  $x$ s as a function of temperature for your chosen value of  $\rho$ . (Plot  $x_0$ ,  $x_1$ , and  $x_2$  on the same graph.) Identify the transition temperatures (half-ionization) for the two ionization stages.

**Solution:** Figures 2-4 show the ionization fraction for three densities. The half-ionization temperatures (defined as the lowest temperature at which the ionization fraction of a species is 0.5) are:

$$\rho = 10^{-4} \text{ g/cm}^3: T(x_1 = 0.5) = 3.2 \times 10^4 \text{ K}, T(x_2 = 0.5) = 8.1 \times 10^4 \text{ K}$$

$$\rho = 10^{-6} \text{ g/cm}^3: T(x_1 = 0.5) = 2.2 \times 10^4 \text{ K}, T(x_2 = 0.5) = 5.4 \times 10^4 \text{ K}$$

$$\rho = 10^{-8} \text{ g/cm}^3: T(x_1 = 0.5) = 1.7 \times 10^4 \text{ K}, T(x_2 = 0.5) = 4.0 \times 10^4 \text{ K}$$

### 3. Stability against convection [10 pts]

- (a) In lecture, we derived the condition

$$\left| \frac{dT}{dr} \right| < \frac{T}{P} \left( 1 - \frac{1}{\gamma_a} \right) \left| \frac{dP}{dr} \right|$$

for stability against convection. Using the appropriate equation(s) of stellar structure and noting the sign of the radial gradients, show that this can be recast as a condition on the luminosity profile:

$$L(r) < \left( 1 - \frac{1}{\gamma_a} \right) \frac{64\pi\sigma_{\text{SB}}T^4GM(r)}{3\kappa_R P}$$

**Solution:** We've derived the condition

$$\frac{\rho}{\gamma P} \frac{dP}{dr} - \frac{d\rho}{dr} > 0$$

for stability against convection. Using the ideal gas law  $P = \rho kT / \mu m_p$ , we can calculate  $d\rho/dr$ , and find

$$\frac{d\rho}{dr} = \frac{\rho}{P} \frac{dP}{dr} - \frac{\rho}{T} \frac{dT}{dr}.$$

Substituting into the condition for stability and simplifying, we obtain:

$$\frac{dT}{dr} > \left( 1 - \frac{1}{\gamma} \right) \frac{T}{P} \frac{dP}{dr}$$

Using the equation of radiative transport,

$$\frac{dT}{dr} = \frac{-3\kappa_R \rho L(r)}{16\pi a c T^3 r^2},$$

and solving the inequality for  $L(r)$ :

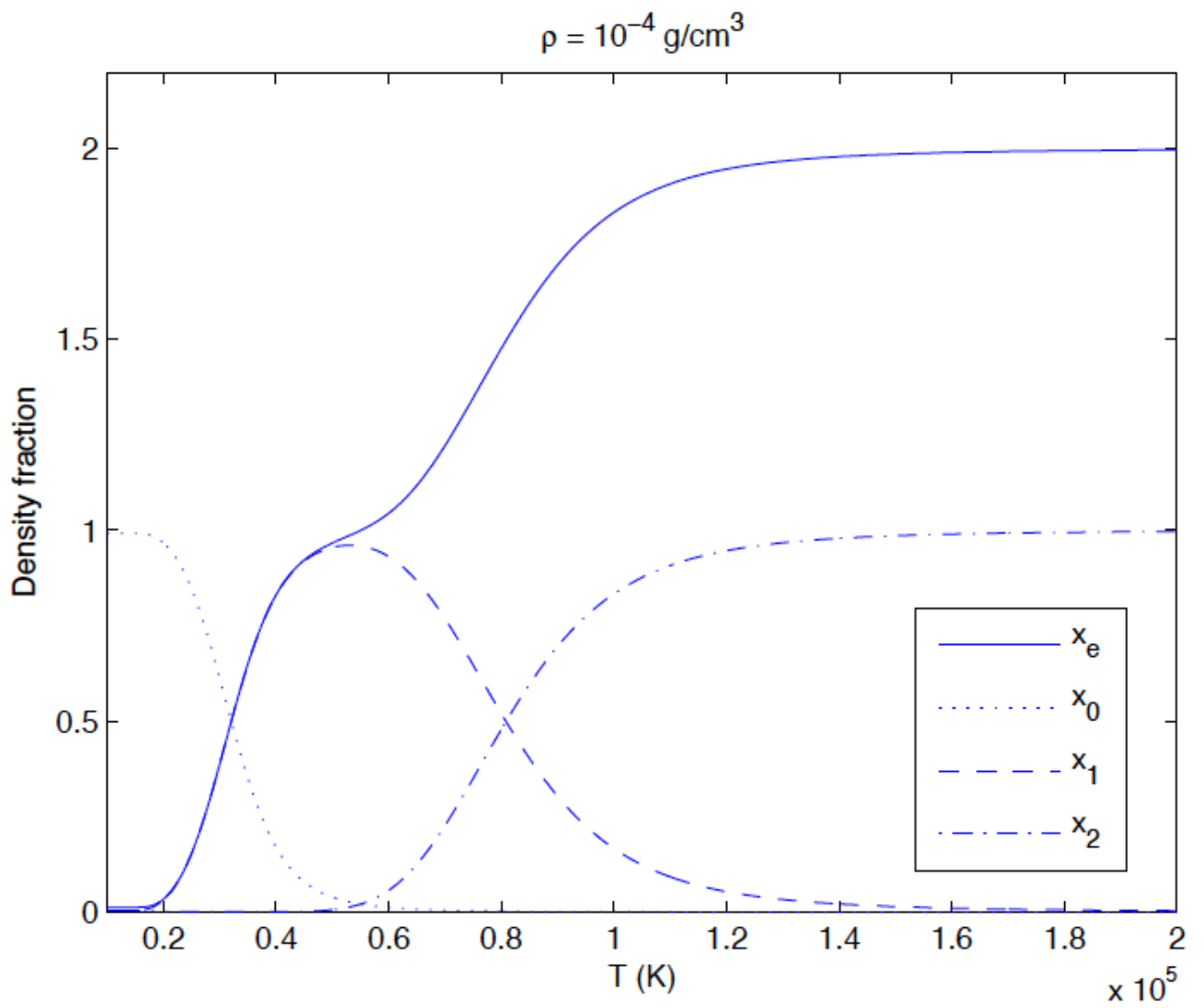


Figure 2: Ionization fractions for pure helium,  $\rho = 10^{-4} \text{ g/cm}^3$ .

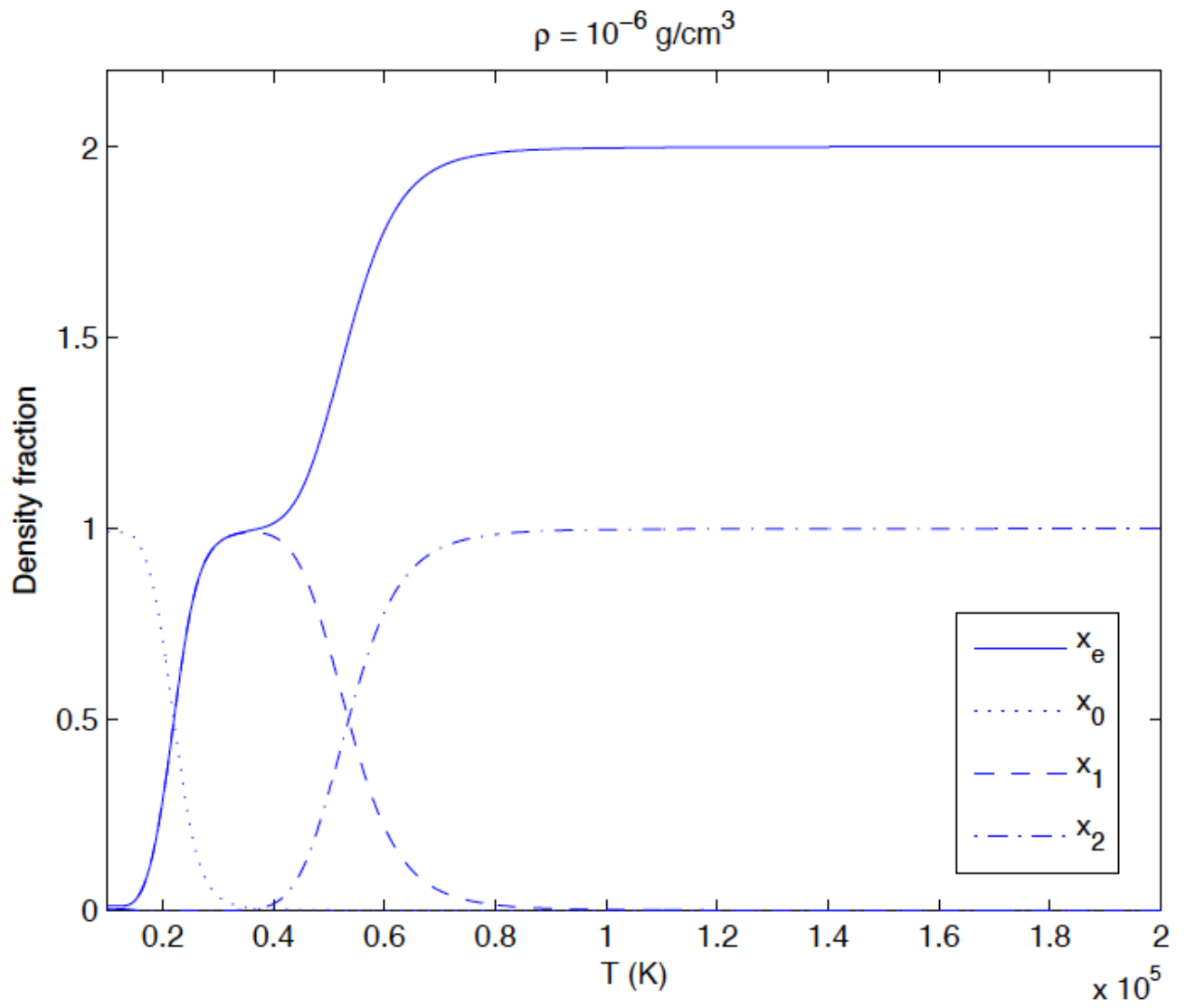


Figure 3: Ionization fractions for pure helium,  $\rho = 10^{-6} \text{ g/cm}^3$ .

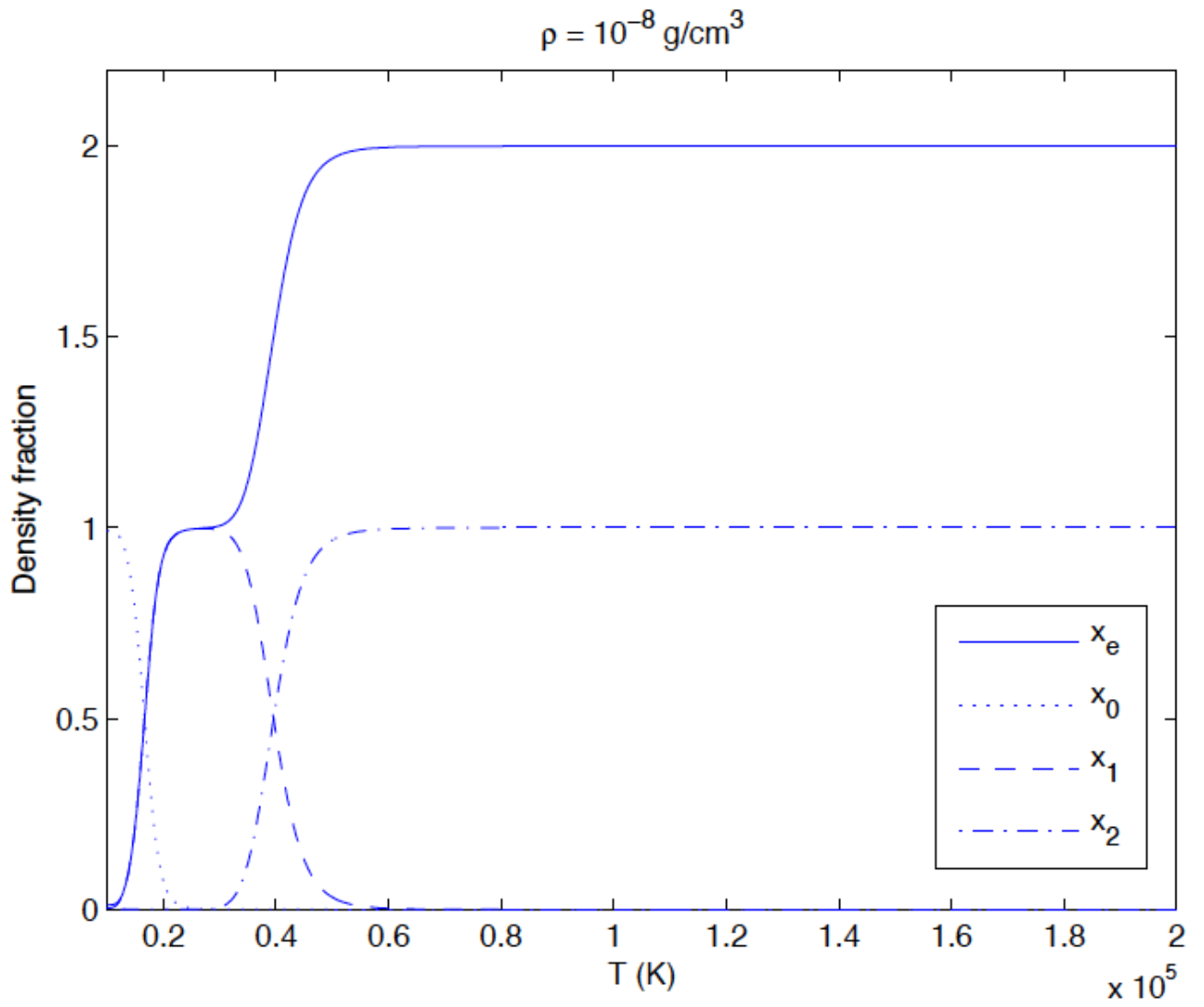


Figure 4: Ionization fractions for pure helium,  $\rho = 10^{-8} \text{ g/cm}^3$ .



$$L(r) < \frac{-16\pi acT^4 r^2}{3\kappa_R \rho P} \frac{dP}{dr} \left(1 - \frac{1}{\gamma}\right)$$

And substituting in the equation for hydrostatic equilibrium, we get

$$L(r) < \left(1 - \frac{1}{\gamma}\right) \frac{16\pi acT^4 GM(r)}{3\kappa_R P}$$

$$L(r) < \left(1 - \frac{1}{\gamma}\right) \frac{64\pi\sigma_{SB}T^4 GM(r)}{3\kappa_R P}$$

- (b) Show that to avoid convection in a stellar region where the equation of state is that of an ideal monatomic gas, the luminosity at a given radius must be limited by

$$L(r) < 1.22 \times 10^{-18} \frac{\mu T^3}{\kappa_R \rho} M(r)$$

where  $\mu$  is the mean molecular weight,  $T(r)$ ,  $\kappa_R$  is the Rosseland mean opacity, and  $M(r)$  is the mass enclosed at radius  $r$ . All quantities are measured in the appropriate cgs units.

**Solution:** For an ideal monatomic gas,  $\gamma = \frac{5}{3}$  and  $P = \rho kT / \mu m_p$ . Plugging in these expressions, we arrive at the desired result (in cgs units):

$$L(r) < 1.22 \times 10^{-18} \frac{\mu T^3}{\kappa_R \rho} M(r)$$

#### 4. Protons or photons? [10 pts]

At the center of the Sun, the density is approximately  $150 \text{ g cm}^{-3}$  and the temperature is about  $15 \times 10^6 \text{ K}$ . Which is larger: the number density of protons, or the number density of photons? Give an order of magnitude estimate of each.

**Solution:** The number density of protons is roughly

$$n_p \approx \frac{\rho c}{m_p} = \boxed{9.03 \times 10^{25} \text{ cm}^{-3}},$$

where we have neglected the effect of He and considered a pure hydrogen composition. The number density of photons is can be related to the temperature

$$n_\gamma = \int \frac{u_\nu}{h\nu} d\nu = \int \frac{4\pi B_\nu(T)}{ch\nu} = \frac{16\pi k_B^3 T^3}{c^3 h^3} \zeta(3) \approx \boxed{6.9 \times 10^{22} \text{ cm}^{-3}},$$

where  $\zeta(3) \approx 1.20$  is the Reimann zeta function. The number density of protons is more than a thousand times higher than the number density of photons.

#### 5. The Eddington limit [15 pts]

A star with sufficiently high radiation pressure will spontaneously eject material from its surface. This sets a practical limit on the maximum luminosity of a star of a given mass.

- (a) [10 pts] Start with the radiative diffusion equation and the equation for hydrostatic equilibrium. Assume the opacity to be frequency-independent, and show that the luminosity at which the radiation pressure gradient equals the hydrostatic pressure gradient is given by

$$L_{\text{Edd}} = \frac{4\pi GMc}{\kappa}, \quad (1)$$

where  $M$  is the stellar mass. This is the ‘‘Eddington luminosity.’’

**Solution:** The second moment of the radiative transfer equation in spherical coordinates is (see problem set 3, problem 8c)

$$c \frac{dP_\nu}{dr} = -\rho \kappa_\nu F_\nu.$$

Integrating over frequency, using the fact that  $\kappa$  is independent of frequency, and using  $F = L/4\pi r^2$ , we find the radiation pressure gradient to be

$$\frac{dP_r}{dr} = -\frac{\rho \kappa L}{4\pi r^2 c}.$$

If radiation pressure is the dominant source of pressure, then, using the equation of hydrostatic equilibrium,

$$-\frac{\rho \kappa L}{4\pi r^2 c} = -\frac{GM}{r^2} \rho.$$

This holds when *all* of the pressure support is provided by radiation. In this case the luminosity no longer depends on density or radius; it is simply

$$L_{\text{Edd}} = \frac{4\pi GMc}{\kappa}.$$

- (b) [5 pts] For ionized hydrogen, a minimum value for  $\kappa$  arises from Thomson scattering, which has cross-section  $\sigma_T = 6.65 \times 10^{-25} \text{ cm}^2$ . Show that for this case

$$L_{\text{Edd}} \approx 3 \times 10^4 L_\odot \left( \frac{M}{M_\odot} \right), \quad (2)$$

where  $L_\odot = 3.839 \times 10^{33} \text{ erg s}^{-1}$  and  $M_\odot = 1.989 \times 10^{33} \text{ g}$ .

**Solution:** Since  $n\sigma = \kappa\rho$ , we have  $\kappa_T = n\sigma_T/\rho$ . Here  $n$  is the number density of scatterers (electrons), but  $\rho$  is the mass density of the medium (mostly protons). From charge neutrality,  $n_e = n_p$ , giving  $\kappa_T = \sigma_T/m_p$ . Plugging this and the solar mass into  $L_{\text{Edd}}$  gives Eq. (2).